

18 BARBITURATE METHODOLOGY	Page 1 of 2
<div>Division of Forensic Science</div> <div>CONTROLLED SUBSTANCES PROCEDURES MANUAL</div>	Amendment Designator:
	Effective Date: 9-December-2003
<div>18 BARBITURATE METHODOLOGY</div> <div> <p>18.1 Brief Pharmacology: Central nervous system depressants commonly known as "Downers"</p> <p>18.2 Drug Group Examples: Butalbital, pentobarbital, secobarbital, allobarbital, amobarbital, butabarbital, barbital, and phenobarbital</p> <p>18.3 Types of Samples:</p> <p>18.3.1 Most barbiturates are found in pharmaceutical preparations.</p> <p>18.4 Scheduling:</p> <ul style="list-style-type: none"> Schedule II Amobarbital, secobarbital, and pentobarbital Schedule III Most barbiturates Schedule IV Phenobarbital Schedule VI or non-controlled Some preparations of phenobarbital, butalbital, and other such barbiturates are specifically exempted from control. Appropriate caution must be exercised when determining their control status. Any questions should be answered by consulting appropriate compendia references such as the PDR, Poison Control, DEA Logo Index and the appropriate state or federal codes, as well as, informing the section supervisor. If any question remains, <u>DO NOT</u> include the schedule in your report. <p>18.5 Extraction: May be extracted from either acidic or weak basic aqueous solutions with organic solvents.</p> <p>18.6 Color Tests Results:</p> <p>18.6.1 Dille-Koppanyi - This is a two part test. Place 2 drops of DK1 reagent in a well. Add sample. Add 1 drop of DK2 reagent. When doing multiple samples, they should be separated to avoid cross-contamination due to reagent spreading. Barbiturates give a purple color. False positives from: glutethimide, theophylline and hydantoins.</p> <p>18.6.2 Co(SCN)₂ - faint blue on barbiturates with an unsaturated side chain (i.e., butalbital).</p> <p>18.6.3 Parri - blue</p> <p>18.7 TLC:</p> <p>18.7.1 Baths: The isopropyl ether bath (TLC7) will separate most of the barbiturates from one another.</p> <p>18.7.2 Detection sprays:</p> <p>18.7.2.1 KMnO₄ reacts with barbiturates with an unsaturated side chain to yield a yellow spot on a purple background.</p> <p>18.7.2.2 HgSO₄ - spray very heavily to give light spots on an off-white background.</p> <p>18.7.2.3 Diphenylcarbazone - overspray for HgSO₄ gives pink spots for barbiturates.</p> <p>18.8 UV:</p> <p>18.8.1 Extraction of the sample may be necessary to get a good spectrum.</p> </div>	

<p align="center">18 BARBITURATE METHODOLOGY</p>	<p align="center">Page 2 of 2</p>
<p align="center">Division of Forensic Science</p> <p align="center">CONTROLLED SUBSTANCES PROCEDURES MANUAL</p>	<p>Amendment Designator:</p>
	<p>Effective Date: 9-December-2003</p>
<div data-bbox="240 260 1495 323"> <p>18.8.2 Barbiturates should be run in both acidic and basic media due to their characteristic bathochromic shift as the solution becomes basic.</p> </div> <div data-bbox="152 354 285 386"> <p>18.9 GC:</p> </div> <div data-bbox="240 417 1409 449"> <p>18.9.1 Extraction or derivatization of the sample may be necessary to get good chromatographic peak shape.</p> </div> <div data-bbox="240 480 943 512"> <p>18.9.2 Alkyl Derivative: trimethylanilinium hydroxide (TMAH)</p> </div> <div data-bbox="334 543 802 575"> <p>18.9.2.1 See GC section 10 for procedure.</p> </div> <div data-bbox="334 606 1409 638"> <p>18.9.2.2 Formation of the methyl derivative will generally decrease the retention time significantly.</p> </div> <div data-bbox="152 669 334 701"> <p>18.10 GC/MS:</p> </div> <div data-bbox="240 732 1247 764"> <p>18.10.1 Barbiturates most often do not exhibit a molecular ion peak and require derivitization.</p> </div> <div data-bbox="152 795 302 827"> <p>18.11 FTIR</p> </div> <div data-bbox="240 858 1003 890"> <p>18.11.1 Extraction may be necessary to obtain a useful FTIR spectrum.</p> </div> <div data-bbox="1484 911 1549 942"> <p align="right">♦ End</p> </div>	